



PENANCE
KEEPING BUSY

BOOK 2 OF THE PENANCE TRILOGY

CR SPENCER

Penance Trilogy

Michael Brown



Penance Trilogy:

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 $2 \text{ NaCl} + \text{F}_2 \rightarrow 2 \text{ NaF} + \text{Cl}_2$
 $2 \text{ H}_2 + \text{O}_2 \rightarrow 2 \text{ H}_2\text{O}$
 $5 \text{ Pb} + 2 \text{ OH}^- + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2 + 2 \text{ Al} + 3 \text{ K}_2\text{SO}_4 + 6 \text{ KBr} + 1 \text{ Al}_2\text{SO}_4 + 3 \text{ H}_2\text{O} + 2 \text{ CO}_2 + 4 \text{ CH}_4$

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 $1 \text{ N}_2 + \text{F}_2 \rightarrow 2 \text{ NF}_3$
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 $3 \text{ HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 $\text{GaBr} + \text{NaBr} + \text{SnO}_4 \rightarrow \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{S}_3 + 5 \text{ NF}_3 + \text{SnF}_2 + \text{N}_2\text{O}_3$ using the equation from problem 2 above answer the following

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