

# HOW TO WRITE BLOG POSTS USING AI

KEEP WRITING EVEN IF YOU'RE SHORT ON TIME OR HAVE WRITER'S BLOCK



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**AW Rasmussen**



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seaborgium determine if the following electron configurations are correct 26 1s<sup>2</sup>2s<sup>2</sup>2p<sup>6</sup>3s<sup>2</sup>3p<sup>6</sup>4s<sup>2</sup>3d<sup>10</sup>4p<sup>6</sup>5s<sup>1</sup> no it should be 3d<sup>10</sup>

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web how many protons neutrons and electrons are in atoms of these isotopes write the complete electron configuration for each isotope answer co has 27 protons 27 electrons and 33 neutrons 1s<sup>2</sup> 2s<sup>2</sup> 2p<sup>6</sup> 3s<sup>2</sup> 3p<sup>6</sup> 4s<sup>2</sup> 3d<sup>7</sup> i has 53 protons 53 electrons and 78 neutrons 1s<sup>2</sup> 2s<sup>2</sup> 2p<sup>6</sup> 3s<sup>2</sup> 3p<sup>6</sup> 3d<sup>10</sup> 4s<sup>2</sup> 4p<sup>6</sup> 4d<sup>10</sup> 5s<sup>2</sup> 5p<sup>5</sup>

**electron configuration gizmo name studocu** - Aug 07 2022

web select the electron configuration tab on show number of electrons introduction beyond argon it is a bit tricky to determine which subshell gets filled next there are several rules that scientists use to determine the electron configurations of larger atoms question how are the electron configurations of elements beyond argon determined