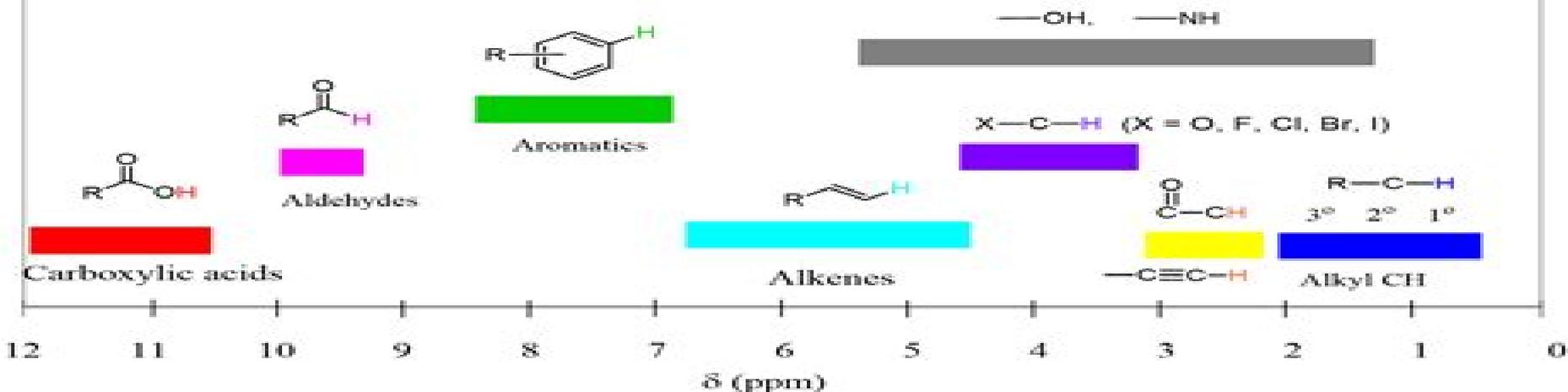


Downfield

Upfield



$R-CH_3$	0.7 - 1.3
$R-CH_2-R$	1.2 - 1.4
R_3CH	1.4 - 1.7
$R-C=C-CH_3$	1.6 - 2.6
$R-C(=O)-CH_3$	2.1 - 2.5
$RO-C(=O)-CH_3$	2.1 - 2.6
$N\equiv C-CH_3$	2.1 - 3.0
$\text{C}_6\text{H}_5-CH_3$	2.3 - 2.7
$R-C\equiv C-H$	1.7 - 2.7
$R-N-CH_3$	2.2 - 2.9
$R-S-CH_3$	2.0 - 3.0

$I-CH_3$	2.0 - 4.0
$Br-CH_3$	2.7 - 4.1
$Cl-CH_3$	3.1 - 4.1
$RO-CH_3$	3.2 - 3.8
$R=H \text{ or alkyl}$	
$R-C(=O)-O-CH_3$	3.5 - 4.8
O_2N-CH_3	4.1 - 4.3
$F-CH_3$	4.2 - 4.8
$R-C=C-H$	4.5 - 6.5
C_6H_5-H	6.5 - 8.0
$R-C(=O)-H$	9.0 - 10

NH and OH peaks are most often broad or may as well be missing completely unless the sample is very dry.

This is also true for any proton capable of making hydrogen bonding:

$R-SH$	1.0 - 5.0
$R-NH_2$ $1^\circ, 2^\circ$	1.0 - 5.0
$R-OH$ $1^\circ, 2^\circ, 3^\circ$	1.0 - 5.0
C_6H_5-OH	4.0 - 7.0
$R-C(=O)-N-R$ $1^\circ, 2^\circ H$	5.0 - 9.0
$R-C(=O)-OH$	11 - 12

Downfield shifts more common

Nmr Data Tables For Organic Compounds

Anthony S. Fauci



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