

ఆమ్లము - క్షారము

Acids

- H^+ అయాన్లు
- కుంఠిత ప్రతిచర్యలు ఉంటాయి
- Blue లిట్మస్ \rightarrow Red లిట్మస్
- వివిధ ఆమ్లాల Indicator ను వివిధ రంగులకు మారుతుంది \rightarrow Red.
- pI (వినియోగం) - No colour

Bases

- OH^- అయాన్లు
- కుంఠిత ప్రతిచర్యలు ఉంటాయి.
- Red లిట్మస్ \rightarrow Blue లిట్మస్
- వివిధ క్షారాలను వివిధ రంగులకు మారుతుంది Yellow
- వినియోగం క్షారాల మారుతుంది - Pink

Note: లిట్మస్ paper colour - పరివర్తన

ఆమ్లత్వాన్ని గుర్తించుట: ① ఒక సమీకరణం 'H' లో మార్పుతో expect $[H_2O]$

- HF - ఫ్లోరోజిక్ ఫ్లోరోజిక్ ఆమ్లం (గాజాపై తగ్గదు రాయదు)
- HCl - హైడ్రోక్లోరిక్ ఆమ్లం
- HI - హైడ్రోజిక్ ఆమ్లం
- H₂S - హైడ్రోజిక్ సల్ఫ్యూరిక్ ఆమ్లం (గాలిలో విత్పంపి ఉంచినప్పుడు colour change)
- H₂SO₄ - సల్ఫ్యూరిక్ ఆమ్లం
- H₂SO₃ - సల్ఫ్యూరిక్ ఆమ్లం
- HNO₃ - నైట్రిక్ ఆమ్లం
- HNO₂ - నైట్రిక్ ఆమ్లం
- H₃PO₄ - ఫాస్ఫోరిక్ ఆమ్లం (Coke drinks)
- H₃PO₃ - ఫాస్ఫోరిక్ ఆమ్లం
- H₃PO₂ - ఫాస్ఫోరిక్ ఆమ్లం
- H₂CO₃ - కార్బోనిక్ ఆమ్లం (నోవోల్)
- HClO₃ - క్లోరిక్ ఆమ్లం
- H₃BO₃ - బోరిక్ ఆమ్లం (త్రిమి సయానైడ్)

② ఒక సమీకరణం 'COOH' లో మార్పుతో వివిధ రంగులు ఉంటాయి.

- H COOH - ఫార్మిక్ ఆమ్లం (వెనిక)
- CH₃COOH - అసిటిక్ ఆమ్లం
- C₂H₅COOH - కార్బోక్సిటిక్ ఆమ్లం

③ ఆమ్లాల విచ్ఛేదనం వివిధంగా గుర్తించుట.
NO₂, N₂O₅, SO₃, P₂O₅, CO₂

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