

# Modern Russian 1

## Workbook

Clayton L. Dawson

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instead of ml we have  $v = \frac{m}{\rho}$  32 g mol<sup>-1</sup> 790 g l<sup>-1</sup> 0.0405 l mol<sup>-1</sup> nonnumber

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web aug 3 2022 volumes of gases and liquids are measured in cubic centimetres  $\text{cm}^3$  or cubic decimetres  $\text{dm}^3$  or litres l 1 mole of any gas at room temperature and pressure occupies  $24 \text{ dm}^3 = 24,000 \text{ cm}^3$  this value is known as the molar volume

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