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# Poetic Bubbles

**L Cohen**



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web write the oxidation and reduction reactions for each redox reaction the first one is done for you 23  $\text{Fe} + \text{CO}_2 \rightarrow \text{FeO} + \text{C}$   
oxidation  $\text{C} + \text{CO}_2 \rightarrow 2\text{CO}$  reduction  $\text{Fe} + 2\text{Fe} \rightarrow 3\text{Fe}$  24  $3\text{Ag} + \text{Ni} \rightarrow 3\text{Ag} + \text{Ni}$  oxidation reduction 25  $\text{Cu} + \text{Pb} \rightarrow \text{Cu} + \text{Pb}$  oxidation reduction 26  $\text{O}_2$

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