

MOLE CONCEPT

Scientific notation
 $N \times 10^n \rightarrow$ Integer
 $N < 10$

Density $\rightarrow \frac{\text{mass}}{\text{volume}}$
 Temperature \rightarrow
 $^{\circ}\text{F} = \frac{9}{5}(^{\circ}\text{C}) + 32$
 $\text{K} = ^{\circ}\text{C} + 273.15$

Limiting Reagent
 The reactant which present in less quantity. gets consumed after sometime. by this we given mole can find LR.

Isotopes - Z same
 A diff
 Isobars - Z diff
 A same
 Isoelectronic - Same num e^s

molar mass - mass of one mole substance in grams.

one mole: - 6.022×10^{23} particles.

percentage composition: - $\frac{\text{mass}_c}{\text{molar mass}_c} \times 100$

Empirical formula = molecular formula

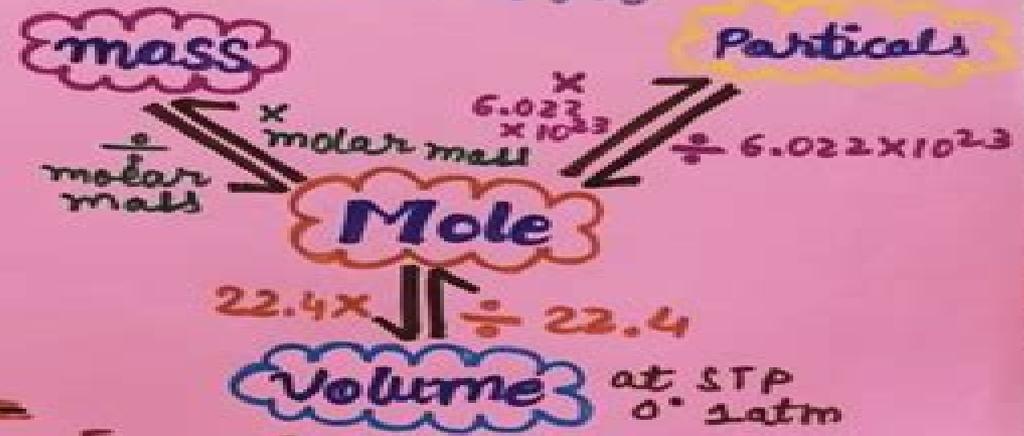
* Mass per cent = $\frac{\text{mass of solute} \times 100}{\text{mass of solution}}$

* Mole fraction = $\frac{\text{No. of moles of A}}{\text{No. of moles of solution}}$

* Molarity (M) = $\frac{\text{No. of mole of solute}}{\text{Volume of solution (lit)}}$

* Molality (m) = $\frac{\text{no. of moles of solute}}{\text{mass of solvent (in kg)}}$

Conversions



Empirical formula - gives simplest whole number ratio of various elements in one molecule of compound.
 formula of $\text{C}_6\text{H}_{12}\text{O}_6$
 Empirical for. CH_2O

Molecular formula - gives actual ratio of atom of various element in one molecule of com.
 $\text{MF} = n \times \text{EF}$
 n is a common factor



Mole Notes

Rosina Ehmann



Mole Notes:

Reviewing **Mole Notes**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Mole Notes**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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