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PROPERTIES OF SATURATED STEAM

Steam Pressure PSIG/ Inches Hg	Saturated Temp °F	Sensible Heat BTU/lb.	Latent Heat BTU/lb.	Total Heat BTU/lb.	Specific Volume (liquid) cu ft/lb	Specific Volume (vapor) cu ft/lb
29.743	32	0	1076	1076	0.01602	3306
20	161	129	1001	1130	0.01640	75.32
10	192	160	982	1143	0.01658	39.21
0	212	180	970	1150	0.01671	26.80
5	227	195	960	1155	0.01682	20.38
10	239	208	952	1160	0.01692	16.49
15	250	218	946	1164	0.01700	13.88
25	267	236	934	1170	0.01714	10.58
50	298	267	912	1179	0.01743	6.69
75	320	290	895	1185	0.01765	4.91
100	338	309	881	1190	0.01785	3.89
125	353	325	869	1193	0.01802	3.23
150	366	338	857	1195	0.01818	2.76
200	388	362	838	1200	0.01847	2.14
250	406	382	821	1202	0.01873	1.74
300	422	399	805	1204	0.01897	1.47

Properties Of Steam

Hugh Longbourne Callendar



Properties Of Steam:

Tables of the Properties of Steam and Other Vapors Cecil Hobart Peabody, 1914 **Thermal Properties of Steam**
George Alfred Goodenough, 1914 Tables of the Properties of Steam and Other Vapors, and Temperature-entropy Table
Cecil Hobart Peabody, 1907 *Thermodynamic properties of steam* Joseph Henry Keenan, Frederick George Keyes, 1953
Thermal Properties of Steam George Alfred Goodenough, 1914 **Tables and Diagrams of the Thermal Properties
of Saturated and Superheated Steam** Lionel Simeon Marks, Harvey Nathaniel Davis, 1919 **Tables of the Properties
of Steam and Other Vapors, and Temperature-Entropy Table (Classic Reprint)** Cecil H. Peabody, 2016-06-26 Excerpt
from *Tables of the Properties of Steam and Other Vapors and Temperature Entropy Table* The *Tables of the Properties of
Steam* were published in 1888 to accompany the author's *Thermodynamics of the Steam Engine* in 1907 they were revised
taking advantage of added information then available and a temperature entropy Table was added to facilitate calculations
for steam turbines The properties of steam have recently been redetermined by new and refined methods that are capable of
great certainty and precision so that computations based upon them Show a satisfactory concordance These tables have been
recomputed with this information and may therefore be used with confidence and may be expected to have permanence The
temperature entropy Table gives solutions of all adiabatic problems and many others both for saturated and for superheated
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Steam and Gas Tables with Computer Equations Thomas F. Jr. Irving,2012-12-02 Steam and Gas Tables with Computer Equations presents tables illustrating the thermodynamic properties of steam and air along with computer equations Additional equations for a number of other gaseous substances which are useful in engineering investigations are included This book is comprised of two chapters and begins with a discussion on the thermodynamic properties of steam which can be divided into saturation and superheat properties The various thermodynamic properties including saturation temperature and pressure and liquid and vapor saturation entropy are represented with three basic types of equations from the triple point to the critical point The accuracy of the properties calculated from the base data is also considered The next chapter deals with the thermodynamic properties of air and other gases ethane hydrogen methane nitrogen oxygen propane n butane including those properties which are useful in engineering design and analysis specific heat at constant pressure and volume enthalpy and entropy function isentropic pressure function etc This monograph will serve as a useful guide for chemists mathematicians and computer programmers and scientists

Properties of Steam and Ammonia (Classic Reprint) G. A. Goodenough,2016-10-08 Excerpt from Properties of Steam and Ammonia The older tables of the properties of steam were neither consistent nor accurate The tabular values were calculated from empirical formulas based chiefly on Regnault's data

and the necessity of consistency was not recognized Two sets of tables have been based on the general theory developed by Callendar These are absolutely consistent but in the light of the knowledge acquired from the Munich experiments they can no longer be regarded as accurate In certain tables that have appeared recently have been embodied the results of the Munich experiments and also the researches of Dr Davis on the total heat of steam These tables are undoubtedly far more accurate than the earlier tables but having a more or less empirical basis they are not rigorously consistent The tables of the properties of saturated and superheated steam here presented are based on a new formulation the essential features of which are discussed in the first section of the book A more complete exposition will be found in Bulletin No 75 Engineering Experiment Station University of Illinois The new theory correlates perfectly the experiments on the volume and specific heat of superheated steam it gives values of the heat content of saturated steam that agree with those deduced by Davis from the throttling experiments and in general it meets satisfactorily all the tests furnished by the available experimental evidence The tables derived from the formulation are necessarily consistent And they are at the same time extremely accurate About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books Find more at www.forgottenbooks.com This book is a reproduction of an important historical work Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections that remain are intentionally left to preserve the state of such historical works

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Steam Tables Joseph H. Keenan, Frederick G. Keyes, Philip G. Hill, Joan G. Moore, 1969-01-16 Steam Tables Thermodynamic Properties of Water Including Vapor Liquid and Solid Phases English Units By Joseph H Keenan M I T Frederick G Keyes M I T Philip G Hill Queen s University and Joan G Moore M I T During the past decade a substantial body of experimental data on

thermodynamic and transport properties of water has been produced and published by research groups in the USSR Great Britain Czechoslovakia Canada and the United States This book presents the results of a new and independent correlation of all this new thermodynamic data and all previously existing data It is a new work to replace the well known and widely used Keenan and Keyes tables The tables in this new book are based upon a unique accomplishment For the first time the whole body of high quality experimental data on liquid and vapor water has been faithfully represented by a single fundamental equation From this equation all thermodynamic properties can be calculated for any state This equation is believed to extrapolate dependably in temperature from the upper limit of precise measurement about 1500 F to about 2400 F Because of the increasing importance to both the practicing engineer and the student of a wide variety of problems that cannot be approximated by steady flow idealization internal energies are tabulated for all states saturated liquid and vapor compressed liquid and superheated vapor A reasonable range of metastable states is covered as extensions of the superheated vapor and compressed liquid tables The Mollier and temperature entropy charts are extended to substantially higher pressures and temperatures This book also includes a table for ice vapor equilibrium an improved chart of isentropic exponents charts of Prandtl number a set of charts of heat capacity of liquid and vapor and extensive tables of viscosity and thermal conductivity reproduced from the documents of the Sixth International Conference on the Properties of Steam The book features legible type set by a computer controlled typesetting machine This results in accuracy compactness and convenience

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