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Oceans Little Charts

RD Boyd



Oceans Little Charts:

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arrhenius model of bases a substance that contains a hydroxide group and dissociates to produce a hydroxide

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they are the same as basic solutions

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nh₂ and s²⁻ are ch₄ nh₃ and hs respectively next we consider that acidity increases with positive charge on the molecule

thus ruling out that s²⁻ is the weakest base

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compounds have characteristics of both acids and bases in the same molecule we will explore the relationship between

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$\text{CaCO}_3 + 2\text{HBr} \rightarrow \text{CaBr}_2 + \text{H}_2\text{O} + \text{CO}_2$ calcium carbonate and hydrobromic acid