

DEPARTMENT 2

John Creasey

The Peril Ahead

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$$I_2 + 2OH^- \rightarrow IO^- + I^- + H_2O$$

$$2J^- \rightarrow J_2 + 2e^-$$

$$Br_2 + 2e^- \rightarrow 2Br^-$$

$$235_{92}U + \frac{1}{0}n \rightarrow 236_{92}U \rightarrow 95A + 140B + E$$

$$Fe_2O_3 + 3H_2 \rightarrow 2Fe + 3H_2O$$

$$2ROH \xrightarrow{H_2SO_4} ROR + H_2O$$

$$2NaBr + H_2SO_4 \rightarrow Na_2SO_4 + 2HBr$$

$$[Fe((CN)_6)]^{3-}$$

$$c(\frac{1}{5}K_2MnO_4) = 1000$$

$$F = \frac{Q_1 Q_2}{4\pi \epsilon_0 r^2}$$

$$TAH = \frac{P_{u2}}{e} = \frac{0.695}{e}$$

$$(z-1)^2 \cdot (\frac{1}{r^2} - \frac{1}{n^2})$$

$$+ \frac{1}{2} O_2 \rightarrow CuO$$

$$CH_3 - C(OH) - CH_3$$

$$PbS + 4H_2O \rightarrow SO_2 + PbO$$

$$Pb^{4+} + 2e^- \rightarrow Pb$$

$$2Zn + 4HNO_3 + 16HCl \rightarrow 2ZnCl_2 + 5Cl_2 + 8H_2O + 2H_2$$

Peril Ahead

Montana. Supreme Court



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