

Write the IUPAC name for the following ions. A correct name includes the charge.

14. Write the formula for the following ions: a) ammonium ion; b) cyanide ion

Chemical Name	Chemical Formula
ammonium ion	NH_4^+
nitrate ion	NO_3^-
acetate ion	$\text{C}_2\text{H}_3\text{O}_2^-$
acetate ion	$\text{C}_2\text{H}_3\text{O}_2^-$
perchlorate ion	ClO_4^-
perchlorate ion	ClO_4^-
perchlorate ion	ClO_4^-

15. Name the following ions: a) cyanide ion; b) nitrate ion

Chemical Name	Chemical Formula
CN^-	cyanide ion
NO_3^-	nitrate ion
Ag^+	silver ion
Cl^-	chloride ion
Mg^{2+}	magnesium ion
Ba^{2+}	barium ion

16. Write the formula for the following ions: a) ammonium ion; b) cyanide ion

Chemical Name	Chemical Formula
ammonium ion	NH_4^+
gold (III) ion	Au^{3+}
nickel (II) ion	Ni^{2+}
lead (IV) ion	Pb^{4+}
tin (II) ion	Sn^{2+}
magnesium (II) ion	Mg^{2+}

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