



## CHEM 33 1<sup>st</sup> LONG EXAM SAMPLEX

1<sup>st</sup> Semester, A.Y. 2022-2023

I. **Modified True or False.** Write TRUE if the statement is correct. If the statement is false, CHANGE the underlined word to make the statement correct.

1. A  $\pi$ -bond is stronger than a  $\sigma$ -bond.
2. The hybridization of both carbon atoms in acetylene is sp.
3. In a reaction, the rate-determining step has the highest activation energy.
4. Iso-pentane and n-pentane are stereoisomers.
5. The boat conformation of cyclohexane is the most stable conformation of the compound.
6. The formation of an internal alkene is thermodynamically favored than that of the terminal alkene.
7. In ozonolysis with oxidative workup, aldehyde by-products are oxidized into alkyl halides.
8. Butyne has higher boiling point than ethyne.
9. In the addition of a hydroxide, the hydrohalogenation of alkynes will follow anti-Markovnikov rule.
10. In IR spectroscopy, higher energy stretching bands occur at lower wave numbers on the  $\text{cm}^{-1}$  scale.

II. **Multiple Choice.** Choose the letter of the best answer.

1. A certain organic compound has a molecular formula of  $\text{C}_7\text{H}_{12}$ . Given this information, the compound cannot be a/an \_\_\_\_\_
  - a. diene
  - b. cycloalkene
  - c. alkyne
  - d. linear straight chain alkane

2. Which of the following structures will yield methylcyclopentane when subjected to catalytic hydrogenation?



d. All of the above

3. Which of the following is true about *cis*-1,2-dimethylcyclopropane and *trans*-1,2-dimethylcyclopropane?
  - a. *cis*-1,2-dimethylcyclopropane is more stable.
  - b. *trans*-1,2-dimethylcyclopropane is more stable.
  - c. Both are equally stable.
  - d. Stability of both compounds cannot be determined.
4. Which kind of organic reactions can reactions A and B be classified, respectively?

# Pkg Acp Chem 33

**Sebastian Brünink**



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