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CHEM 201, Section 14
September 14, 2022

CHEM 201: EXPERIMENT #2

Analysis of Phosphate Ion

Procedure:

Refer to the per-lab on pages 3-4 of the lab notebook for an overview of the procedure of this experiment. There were no changes made to the procedure. The concentration of the fertilizer solution was 2.80 g/L and was composed of 15% Nitrogen, 15% Phosphorus, and 15% Potassium. The energy drink used for the subsequent part of the experiment was Coke. The volume of the bottle of Coke was 200mL. Both the fertilizer and energy drink were 1:30 dilutions.

Experimental Data:

PART A- Calibration of Phosphate

Wavelength λ (nm)	375	385	395	405	415
Absorbance (A)	0.724	0.507	0.384	0.307	0.241

Wavelength with max absorbance (λ_t): 375nm

Absorbance of Ammonium Vanadomolybdate at max wavelength (A_r): 0.167

PART B - Fertilizer

1:30 dilution

Sample:	Wavelength λ (nm):	Absorbance of Fertilizer (A_t):
1	375	0.677
2	375	0.615
3	375	0.604

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