



Name	Lenard Andrew D. Birondo		Date Performed:
Course & Year			Date Submitted:

OBJECTIVES

1. Learn which conditions allow for faster evaporation.
2. Understand the different factors that affect the evaporation of a liquid, such as temperature, surface area, and air flow.
3. Distinguish between evaporation and boiling.
4. Comprehend the relationship between the boiling point temperature of a liquid and its rate of evaporation.
5. Recognize the significance of knowing the rate of evaporation of common chemical liquids.
6. Emphasize the importance of personal protective equipment (PPE) when handling volatile substances to prevent adverse health effects.

THEORY

Evaporation is the process where a substance transitions from its liquid to gaseous state below its boiling temperature (Britannica, 2024). Unlike boiling, which occurs at a specific temperature, evaporation happens gradually as molecules gain enough energy to escape the liquid surface. Factors influencing evaporation include temperature, wind speed, surface area, and humidity (AccessScience, 2015; Byju's, 2022). Higher temperatures increase evaporation rates by speeding up molecule movement, while wind and larger surface areas enhance evaporation by carrying away water vapor and exposing more liquid to air. Lower humidity levels also promote faster evaporation, as drier air can absorb more vapor (USGS, 2019).

Boiling, in contrast to evaporation, is a rapid process characterized by the formation of bubbles within the liquid when it reaches its boiling point (Bobrowsky, 2020). It requires external heat and occurs uniformly throughout the liquid. Substances with higher boiling points generally evaporate more slowly than those with lower boiling points (Stefan V., 2015). In addition, as a liquid is heated, the average kinetic energy of its particles increases. Hence, the rate of evaporation increases as more and more molecules are able to escape the liquid's surface into the vapor phase (LibreText, 2022). Both processes evaporation and

THEORY (Continued)

boiling play crucial roles in environmental and industrial contexts, influencing weather patterns, water cycles, and various chemical processes.

Pkgacp Che 2203 Lab Manual

Marco Cascella



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