

3. Cannizzaro Reaction

- Aldehydes which do not have an α -hydrogen atom can give this reaction
- Aldehyde undergo self oxidation and self-reduction (disproportionation) reaction on heating with concentrated alkali.
- In this reaction, one molecule of the aldehyde is reduced to alcohol while another is oxidised to carboxylic acid salt



- When Cannizzaro reaction is carried out between two different aldehydes, it is called cross Cannizzaro reaction.
- When one of the aldehyde is formaldehyde then it will always be oxidized.



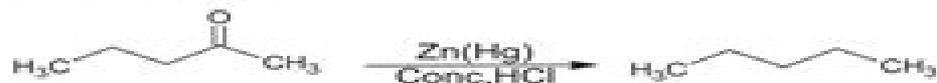
4. Carbylamine Reaction

- Aliphatic and aromatic primary amines on heating with chloroform and ethanolic potassium hydroxide form isocyanides or carbylamines which are foul smelling substances.
- Secondary and tertiary amines do not show this reaction.
- This reaction is also known as reaction isocyanide test
- used as a test for primary amines



5. Clemmensen Reduction

- The carbonyl group of aldehydes and ketones is reduced to CH₂ group (Conversion of Carbonyl compounds to alkanes) on treatment with zinc amalgam and concentrated hydrochloric acid



6. Coupling Reaction

- The azo products obtained have an extended conjugate system having both the aromatic rings joined through the -N=N- bond.
- These compounds are often coloured and are used as dyes.
- Example of electrophilic substitution reaction.
- Benzene diazonium chloride reacts with phenol in which the phenol molecule at its para position is coupled with the diazonium salt to form p-hydroxyazobenzene.



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