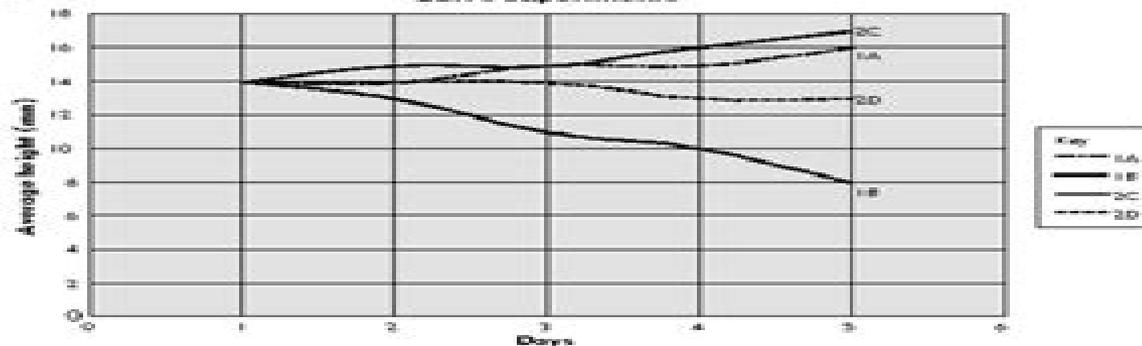


Carl's new experiments

- Qualitative. They were qualitative because all observations were recorded using only words.
- Carl thought that he should also do a quantitative analysis; that is, record the observations using numbers to measure how much the grass actually grows under different situations.
- Controls.
 - Every experiment needs a control against which results can be compared.
- Types of grass, amount of grass.
- The amount of water was kept the same.
- The amount of sunlight.
- The amount of water.
- The soil would have had some water in it on Day 1 and would take a few days to dry out.
- Measurements to be taken at the same time of day. Same ruler to be used and measured from the same place. Others as appropriate.
- Repeating measurements is a good way to improve accuracy. Once a collection of different measurements is taken, an average can be obtained.

11 a Carl's experiments



- 1A 16 or 17 mm
 - 1B 6 or 7 mm
 - 2C 16 or 17 mm
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Liyong Dong



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web may 18 2021 the elements on the periodic table are listed in order of ascending atomic number the periodic table has a special shape that will become important to us when we consider the organization of electrons in atoms chapter 8 one immediate use of the periodic table helps us identify metals and nonmetals

gcse chemistry revision topic 1 atomic structure and the periodic - Mar 30 2022

web dec 4 2020 this is an excellent revision resource for topic 1 atomic structure and the periodic table of gcse chemistry double science that has taken me many weeks to make for my students i call this the must know questions more importantly

they come with answers which is the part that many resources lack but not this one

the periodic table aqa gcse chemistry combined science - Apr 30 2022

web use the data sheet to answer this question the electronic structures of some atoms and ions are shown in figure 1 below figure 1 which structure represents an atom of sodium

understanding atomic structure and periodicity at post 16 rsc - Jun 01 2022

web nov 8 2023 assess students understanding of prior learning by revisiting the topic of atomic structure display this poster in your classroom before tasking 14 16 year old learners to draw electron configuration diagrams using the resource s easy to follow steps do some follow up some work with your students on the reactivity trends in groups 1 and 7

atomic structure the periodic table 3 1 1 save my exams - Nov 06 2022

web elements are arranged on the periodic table in order of increasing atomic number where each element has one proton more than the element preceding it hydrogen has 1 proton helium has 2 protons lithium has 3 etc the table is arranged in vertical columns called groups numbered i viii and in rows called periods elements in the same group have

atomic structure practice khan academy - Aug 03 2022

web atomic structure google classroom what three particles make up an atom choose 1 answer protons neutrons and photons a protons neutrons and photons positrons neutrons and electrons b positrons neutrons and electrons protons electrons and positrons c protons electrons and positrons protons neutrons and electrons d