

## SEMI MICRO QUALITATIVE ANALYSIS OF SIMPLE INORGANIC SALT

<b>I. PRELIMINARY TEST</b>			
	EXPERIMENT	OBSERVATION	INFERENCE
1.	Colour and appearance :		
2.	Solubility: A pinch of the salt is shaken with $\frac{3}{4}$ test tube of water	a. soluble  b. insoluble	a. S.S. prepared in water  b. S.S prepared in dil.HCl
<b>II. IDENTIFICATION OF ACID RADICAL</b>			
<b>Reactions of group-I acid radical</b>			
	EXPERIMENT	OBSERVATION	INFERENCE
3.	Salt + dil. $H_2SO_4$  Pass the gas through lime water.	Brisk effervescence , a colorless gas is evolved.  Lime water turns milky.	Carbonate ion ( $CO_3^{-2}$ ) is present and confirmed
<b>Reactions of group-II acid radical</b>			
4.	Salt + Conc. $H_2SO_4$ in a dry test tube.	a. vigorous reaction , a colorless gas which gives dense white fumes with a glass rod dipped $NH_4OH$ shown near the mouth of the T.T b. reddish brown vapours are evolved	a. Chloride ion ( $Cl^-$ ) is present  b. Bromide ion ( $Br^-$ ) is present

# Semi Micro Qualitative Analysis

**Padhraic Smyth**



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