

Matter and Materials

Grade 7 Chemical Science
Study Guide – Year 8

Physical Properties of Materials

- Strength: Ability to withstand forces of pressure without breaking.
- Flexibility: Ability to bend without breaking.
- Softening and Melting Points:
 - Softening Point: Temperature at which a liquid form starts to get.
 - Melting Point: Temperature at which a solid form starts to melt.
- Chemical Resistance: Ability of a material to allow another material to pass through it.
- Heat Resistance: Ability to prevent heat from being lost.



Working with Matter

- Materials are chosen based on their properties for specific uses.
- Additional factors such as cost, safety, and factors like giving a material mechanical strength.

Structure

- It is a characteristic feature of some substances with a fixed physical structure.
- Components of a mixture are in separated form in their physical structure.



Impact on the Environment

- Production and use of materials has negative, positive, and both of the environment.
- Consideration include air quality, resource depletion, and energy requirements.

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Methods of Separation

Separating Funnel

Materials to be separated (e.g., oil and water)

To separate solids from liquids

Using magnetic materials (e.g., iron from sand)

To separate a mixture of a solid from a liquid

Distillation (separating mixtures of liquids)

Using a filter to separate a solid from a liquid



Properties

Chemical Properties

Physical Properties

Separation

Chemical Properties

Physical Properties

Quality of Materials

- The factors include factors like purity, cost, use, and time.
- Testing helps in identifying which material from the set of substances are up to the mark.



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States of Matter and Solids

Solids

- Regular arrangement of particles.
- Incompressible (cannot be compressed).



Liquids

- Particles are close together.
- Can be compressed (slightly).
- Incompressible (cannot be compressed).



Investigating Solids, Liquids, and Gases

Volume

Mass

Temperature

Pressure

Volume

Mass

Temperature

Pressure

The solid and the liquid expand to fill the container.

- The solid and the liquid expand to fill the container.
- The liquid and the gas expand to fill the container.
- The gas and the liquid expand to fill the container.

The particles are close together and cannot move past each other.

The particles are close together and can move past each other.

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States of Matter and Solids

Solids

Regular arrangement of particles and cannot be compressed.



Liquids

Particles are close together and can move past each other.



Expansion

When a substance is heated, it expands. This is because the particles in the substance are vibrating more rapidly and pushing each other further apart.

Health and Safety

Chemical

Physical

Biological

Environmental

Health

Safety

Health

Safety



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