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web solution a titrant volume 0 ml the solution pH is due to the acid ionization of HCl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0 100 M the pH of the solution is then $pH = \log 0 100 = 1 000$ $pH = \log 0 100 = 1 000$ b titrant volume 12 50 ml

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a q naoh a q h 2 o l nacl a q

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