

# Bittersweet *Revenge*



# Revenge Is Bittersweet

**RJ Shavelson**



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number an unchanging constant whether you are counting hydrogen atoms carbon dioxide molecules or carbonate ions a mole is always  $6.023 \times 10^{23}$  but as different particles have different mass their molar mass varies

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web zero degrees celsius or 273 k is defined as standard temperature and one atmosphere or 760 mm hg is defined as standard pressure the volume occupied by one mole of a gas is called molar volume therefore the molar volume of a gas at standard temperature and pressure stp is 22.4 l

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