

Acids, bases and salts

Lesson 4: Preparation of salts



Grade 10
Mrs D.N.Soopahul

Solubility rules

- All Nitrates are **soluble**
- All sulfates are **soluble** except \rightarrow BaSO_4 , CaSO_4 , PbSO_4
- All Chlorides are **soluble** except \rightarrow PbCl_2 (soluble in hot water), AgCl , HgCl_2
- Potassium, Sodium, Ammonium salts are soluble
- All carbonates are **insoluble** except K_2CO_3 , Na_2CO_3

Salt

Hydrated salt: A salt that contains water of crystallisation E.g hydrated Copper(II) Sulfate is **blue** in colour and has formula $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

Note that the **5 H_2O** represent the water of crystallisation in the salt

When a hydrated salt is heated, it loses its water of crystallisation and becomes an **anhydrous salt**.

Salt Lesson

Joacim Rocklöv



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