

Phase Transformation - Basic Concepts

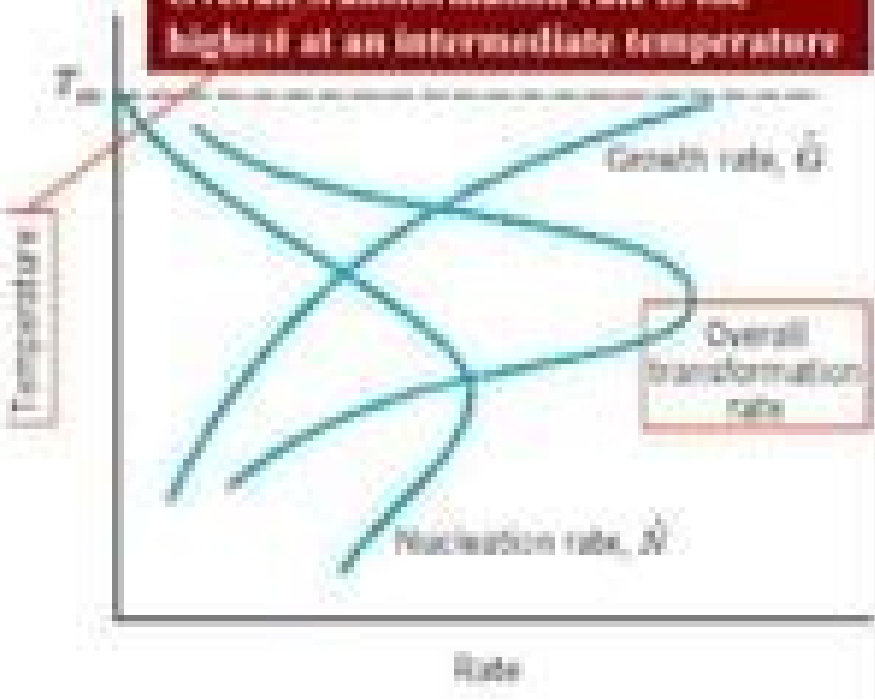
- Growth
 - through diffusion
 - growth rate

$$\dot{G} \propto \exp\left(-\frac{Q}{kT}\right)$$

Temperature is often the most important and used to induce and control a phase transformation

Q: activation energy (often $Q = Q_a$)

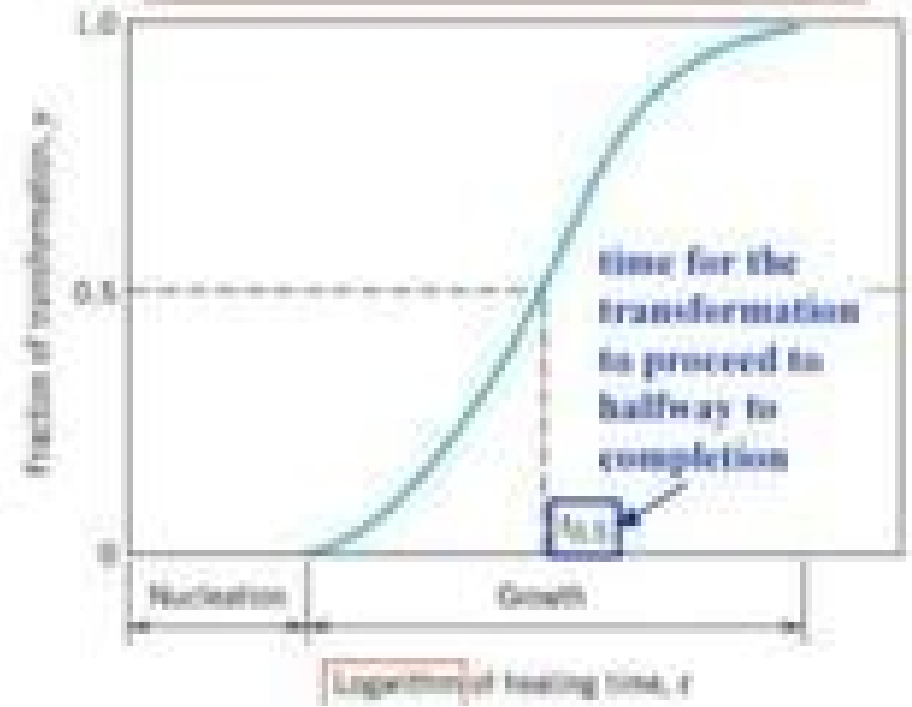
Overall transformation rate is the highest at an intermediate temperature



- Fraction of transformation, y
 - how much of the parent phase has been transformed
 - for solid state transformation at a certain temperature

$$y = 1 - \exp(-kt^n)$$

Rate of transformation : $r = \frac{1}{t_{0.5}}$



Phase Transformations In Solids Symp

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